

www.elsevier.nl/locate/carres

Carbohydrate Research 321 (1999) 1-3

Rapid Communication

InCl₃-induced C-glycosylation of per-O-acetylglycals with allyltrimethylsilane

Rina Ghosh a,*, Debasish De a, Biswajit Shown a, Swaraj B. Maiti b

^a Department of Chemistry, Jadavpur University, Calcutta 700 032, India

Received 1 June 1999; accepted 13 August 1999

Abstract

InCl₃ has been used for the first time as a mediator for the C-glycosylation of per-O-acetylglycals with allyltrimethylsilane, and the reactions proceeded to give products in good-to-excellent yield and in high diastereo-selectivity. The 1,5-anti diastereo-selectivity in the allylation of 3,4,6-tri-O-acetyl-1,5-anhydro-2-deoxy-D-arabino-hex-1-enitol and 3,4-di-O-acetyl-1,5-anhydro-2,6-dideoxy-L-arabino-hex-1-enitol and the 1,4-anti diastereo-selectivity with 3,4-di-O-acetyl-1,5-anhydro-2-deoxy-D-arabino-pent-1-enitol are in the range 80–90%, whereas 3,4,6-tri-O-acetyl-1,5-anhydro-2-deoxy-D-arabino-hex-1-enitol furnished exclusively the corresponding arabino-pendactopyranosyl compound. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: InCl₃; C-Glycosylation; Allyltrimethylsilane; Glycal

C-Glycosylation is a key reaction for the introduction of a carbon chain into sugars. Due to the importance [1] of C-glycosylic compounds (often termed 'C-glycosides'), the present subject has received considerable attention in recent years [2]. Allyl C-glycosides are particularly attractive due to the presence of the terminal double bond that is amenable to easy functionalisation leading to other chiral molecules [3], as well as to carbon analogues of synthetic carbohydrate vaccines.

Previous reports on allyl 2,3-unsaturated per-O-acetyl C-glycosides include synthesis via carbon–Ferrier rearrangement [4a,b] from substituted D-glycals [5a,b,c] or from unprotected glycals [5d] and allyltrimethylsilane at low temperatures (-50 to -78 °C) in the

presence of BF₃·OEt₂, TiCl₄, TiF₄, SnF₄ and Me₃SiOTf. Although InCl₃ has been extensively investigated for a number of C-C bond formation reactions, such as the coupling of allylstannanes [6a,b] and aldehydes, Diels-Alder reactions [6c,d], water-mediated Mukaiyama aldol reactions [6e-h], Michael reactions [6i,j], aldol-type Mannich reactions [6k] etc., to the best of our knowledge its use as a mediator in allylsilane reactions, particularly in the carbohydrate field, has not been reported. This fact gave us impetus to study the carbon-Ferrier rearrangement of per-Oacetylglycals with allyltrimethylsilane utilising InCl₃ as the mediator (Scheme 1).

Our general reaction conditions involve the addition of allyltrimethylsilane (2 equivalents) to a stirred suspension of per-O-acetylglycal (1 equivalent) and InCl₃ (1.5–2 equivalents) in

^b Department of Chemistry, Bangabasi College, Calcutta 700 009, India

^{*} Corresponding author. Tel.: +91-473-4044.

Table 1 InCl₃-induced C-glycosylation of per-O-acetylglycals with allyltrimethylsilane

Substrate	Product a,b	Time (h)	Yield (α/β)	
AcO management of the second o	OAc O 2a	1.5	95% (9:1)	
AcO OAc	OAc O	24	78% (100:0)	
AcO OAc	AcO 2c	1.5	100% (18:1)	
AcO of OAc	AcO Zd	2	70% (9.5:1)	

^a For characterisation data for 2a, 2b, and 2d, see Ref. [8].

dichloromethane at room temperature (20–30 °C). After completion of the reaction as determined by thin-layer chromatography (TLC), the reaction was quenched with saturated aqueous sodium bicarbonate, diluted with dichloromethane, and worked up in the usual manner. The crude product was purified by chromatography on silica gel to afford the allyl 2,3-unsaturated acetyl *C*-glycoside in high yield and with high α-diastereoselectivity (see Table 1). All products were characterised by ¹H and ¹³C NMR spectroscopy (at 300 and 75 MHz, respectively), by elemental analysis,

Scheme 1.

and by comparing the $[\alpha]_D$ values with those in the literature. The α : β ratios were determined by GLC.

The yield and 1,5-anti diastereoselectivity in the allylation of 3,4,6-tri-O-acetyl-1,5-anhydro-2-deoxy-D-arabino-hex-1-enitol (1a) (95%, α : β 9:1) and 3,4-di-O-acetyl-1,5-anhydro-2,6dideoxy-L-arabino-hex-1-enitol (1c) (100%, α:β 18:1) are quite high. The 1,4-anti diastereoselectivity with 3,4-di-O-acetyl-1,5-anhydro-2deoxy-D-erythro-pent-1-enitol (1d) followed the same trend as also observed earlier by Hosokawa et al. [7] in the C-glycosylation of allyltrimethylsilane and silyl acetylenes with acetylated pentopyranoglycals induced by other Lewis acids. 3,4,6-Tri-O-acetyl-1,5-anhydro-2-deoxy-D-lyxo-hex-1-enitol (1b), however, gave exclusively the corresponding C- α -galactoside, i.e., 3-(4,6-di-O-acetyl-2,3dideoxy-α-D-threo-hex-2-enopyranosyl)-1-pro-

^{b 1}H NMR data (CDCl₃) for compound **2c**: δ 1.24 (d, 3 H, J 6.5 Hz, CH₃), 2.08 (s, 3 H, OCOCH₃), 2.33 (m, 1 H, CH₂CH=CH₂), 2.44 (m, 1 H, CH₂CH=CH₂), 3.90–3.94 (dq, 1 H, J 6.5 Hz, 4.7 Hz, H-5'), 4.17–4.23 (m, 1 H), 4.81–4.90 (m, 1 H), 5.08–5.16 (m, 2 H), 5.75–5.95 (m, 3 H).

pene (**2b**) (no β compound could be detected either by GLC or by NMR spectroscopy). Repeating the reactions with a catalytic amount of InCl₃ furnished unchanged starting per-O-acetylglycals in all the above-mentioned cases.

It may be mentioned that reactions of 1a with trimethylphenylsilane and ethynyltrimethylsilane in the presence of $InCl_3$ (even with > 2 equivalents) failed to give the desired C-glycoside product, as was noted earlier by Isobe and co-workers [5a] on similar reactions induced by $BF_3 \cdot OEt_2$.

In conclusion, the present method of InCl₃-induced C–C bond formation provides a highly efficient alternative to the existing methodologies for the synthesis of allyl 2,3-unsaturated-C-α-glycosides. The process has several notable advantages, such as operational simplicity and mild reaction conditions that make use of a reagent of relatively low toxicity. Moreover, this is the first demonstration of InCl₃-mediated reaction of allylsilane with glycals.

Acknowledgements

Financial support to R.G. (01/1463/97/EMR-II) and D.D. (JRF) from CSIR, New Delhi, is gratefully acknowledged. The authors also thank Dr A.K. Sen, Jr. (IICB, Calcutta) for GLC work.

References

(a) M.D. Lewis, J.K. Cha, Y. Kishi, J. Am. Chem. Soc.,
104 (1982) 4976–4978. (b) L. Paterson, L.E. Keown,
Tetrahedron Lett., 38 (1997) 5727–5730. (c) A.B. Smith
III, L. Zhuang, C.S. Brook, A.M. Boldi, M.D. McBriar
W.H. Moser, N. Murase, K. Nakayama, P.R. Verhoest,

- Q. Lin, Tetrahedron Lett., 38 (1997) 8667–8678. (d) K. Horita, Y. Sakkurai, M. Nagasawa, S. Hachiya, O. Yonemitsu, Synlett, (1994) 43–45. (e) R.V. Weatherman, K.H. Mortell, M. Chervenak, L.L. Kiessling, E.J. Toone, Biochemistry, 35 (1996) 3619–3624. (f) R.V. Weatherman, L.L. Kiessling, J. Org. Chem., 61 (1996) 534–538. (g) D.P. Sutherlin, T.M. Stark, R. Hughes, R.W. Armstrong, J. Org. Chem., 61 (1996) 8350–8354.
- [2] (a) Y. Du, R.J. Linhardt, Tetrahedron, 54 (1998) 9913–9959. (b) M.H.D. Postema, C-Glycoside Synthesis, CRC Press, Boca Raton, 1995. (c) D.E. Levy, C. Tang, The Chemistry of C-Glycosides, Pergamon, Oxford, 1995. (d) P. Sinaÿ, Pure Appl. Chem., 69 (1997) 459–463. (e) J.-M. Beau, T. Gallagher, Top. Curr. Chem., 187 (1997) 1–54. (f) F. Nicotra, Top. Curr. Chem., 187 (1997) 55–83.
- [3] S. Hanessian, Total Synthesis of Natural Products: The Chiron Approach, Pergamon Press, Oxford, 1984.
- [4] (a) R.J. Ferrier, N. Prasad, J. Chem. Soc. C, (1969) 570–575. (b) S.J. Danishefsky, S. De Ninno, P. Lartey, J. Am. Chem. Soc., 109 (1987) 2082–2089.
- [5] (a) Y. Ichikawa, M. Isobe, M. Konobe, T. Goto, Carbohydr. Res., 171 (1987) 193–199. (b) S.J. Danishefsky, J.F. Kerwin, Jr., J. Org. Chem., 47 (1982) 3803–3805. (c) M. Isobe, R. Nishizawa, S. Hosokawa, T. Nishikawa, Chem. Commun. (Cambridge), (1998) 2665–2666. (d) K. Toshima, G. Matsuo, T. Ishizuka, Y. Ushiki, M. Nakata, S. Matsumura, J. Org. Chem., 63 (1998) 2307–2313.
- [6] (a) M. Yasuda, T. Miyai, I. Shibata, A. Baba, R. Nomura, H. Matsuda, Tetrahedron Lett., 36 (1995) 9497-9500. (b) D. Behnke, S. Hamm, L. Hennig, P. Welzel., Tetrahedron Lett., 38 (1997) 7059-7062. (c) G. Babu, P. Perumal, Tetrahedron Lett., 38 (1997) 5025–5026. (d) T.-P. Loh, J. Pei, M. Lin, Chem. Commun. (Cambridge), (1996) 2315-2316. (e) T.-P. Loh, J. Pei, G.-Q. Cao, Chem. Commun. (Cambridge), (1996) 1819-1820. (f) T.-P. Loh, J. Pei, K. Siong-Vekoh, G.-Q. Cao, X.-R. Li, Tetrahedron Lett., 38 (1997) 3465-3468 and 3993. (g) T.-P. Loh, G.-L. Chua, J.J. Vittal, M.-W. Wong, Chem. Commun. (Cambridge), (1998) 861-862. (h) K. Kobayashi, T. Busujima, S. Nagayama, Tetrahedron Lett., 39 (1998) 1579-1582. (i) T.-P. Loh, L.-L. Wei, Synlett, (1998) 975-976. (j) T.-P. Loh, L.-L.Wei, Tetrahedron, 54 (1998) 7615-7624. (k) T.-P. Loh, L.-L.Wei, Tetrahedron Lett., 39 (1998) 323-
- [7] S. Hosokawa, B. Kirschbaum, M. Isobe, *Tetrahedron Lett.*, 39 (1998) 1917–1920.
- [8] For preparation of per-O-acetylglycals see: B.K. Shull, Z. Wu, M. Koreeda, J. Carbohydr. Chem., 15 (1996) 955–964; for [α]_D and the ¹H NMR spectrum of compound 2b see, J.L. Macro-Contelles, C. Farnández, A. Gómez, N. Martin-León, Tetrahedron Lett., 31 (1990) 1467–1470, and for compounds 2a and 2d see Refs. [5a] and [7], respectively.